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## Attestation

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Patentanmeldung Nr. Patent application No. Demande de brevet n°

03405467.6

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For the President of the European Patent Office

Le Président de l'Office européen des brevets  
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Ciba Specialty Chemicals Holding Inc.  
Klybeckstrasse 141  
4057 Basel  
SUISSE

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(Falls die Bezeichnung der Erfindung nicht angegeben ist, siehe Beschreibung.  
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Aqueous liquid composition of cyclodextrine or cyclodextrine derivatives and a process using the said composition

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Aqueous liquid compositions of cyclodextrine or cyclodextrine derivatives and a process using the said composition

The present invention relates to an aqueous liquid composition of excellent stability in finishing applications comprising cyclodextrin or a derivative thereof, a resin finishing agent 5 or a crosslinking agent and at least one emulsifier, and a process for the treatment of suitable substrates, in particular fibre materials, using the said composition.

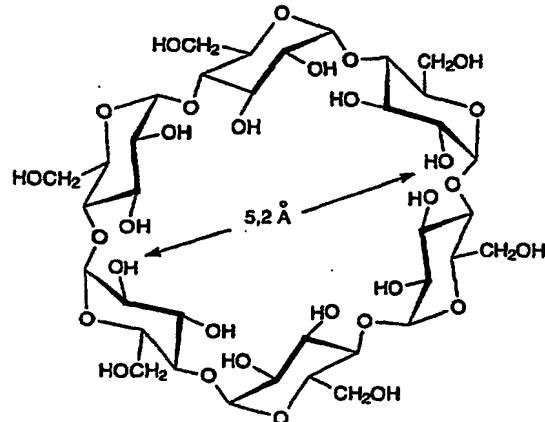
Industrial finishing processes are usually carried out in an aqueous medium. When used for the finishing process finishing agents commercially available in the form of powder or granule 10 must be dissolved in an aqueous medium using water.

In recent years, such finishing processes have been mechanized and automated in many aspects, and therefore the finishing agents have been eagerly required to be made into a form suitable for a dispensing system.

15

Cyclodextrins are cage like molecules of a cyclic configuration made up of a varying number of D-glucopyranosyl units, such as 6, 7 or 8 units ( $\alpha$ -,  $\beta$ - or  $\gamma$ -cyclodextrins), connected by alpha-(1,4) glycosidic linkages, thereby defining a central cavity. The chemical formula of  $\alpha$ -cyclodextrin is depicted below.

20



The natural cyclodextrins are produced from starch by the action of cyclodextrin glycosyltransferase (CGTase), an enzyme produced by several organisms, *Bacillus*

macerans being the earliest source. The most stable three-dimensional molecular configuration for these cyclic oligosaccharides takes the form of a toroid with the upper (larger) and lower (smaller) opening of the toroid presenting secondary and primary hydroxyl groups, respectively, to the solvent environment. The interior of the toroid is hydrophobic as

- 5 a result of the electron rich environment provided in large part by the glycosidic oxygen atoms. It is the interplay of atomic (Van der Waals), thermodynamic (hydrogen bonding), and solvent (hydrophobic) forces that accounts for stable complexes that may be formed with chemical substances while in the apolar environment of the cyclodextrin cavity. The complex exists in an equilibrium dependent upon the concentrations of the cyclodextrin, the guest  
10 molecule and water. The rate at which the associated complex is formed is determined in large part by the accessibility of the guest molecule to the cyclodextrin cavity and the magnitude of the net thermodynamic driving force.

15 Uncomplexed cyclodextrin derivatives have been used as finishing agents for the treatment of fibre materials in order to reduce or prevent malodors due to perspiration (DE-A-40 35 378). Moreover, uncomplexed cyclodextrin derivatives allow for the complexation of fragrances and perfumes (DE-A-40 35 378) or antimicrobial substances (WO-A-02/22941), which are released slowly and impart long-lasting fragrance or a prolonged antimicrobial effect to the finished textile material. The prolonged presence of antimicrobials makes the  
20 substrates more hygienic, less prone to cross contamination and fresher.

An aqueous liquid composition of a cyclodextrine or a cyclodextrine derivative thereof is considered advantageous, since it is suitable for the automatic weighing and dispensing system and causes no powder-scattering during handling, resulting in no pollution of the  
25 working environment, and moreover it can serve saving energy and labor, since substituted cyclodextrin derivatives are prone to swelling upon contact with water and therefore have a strong tendency to form lumps which can hardly be dissolved.

Accordingly it is the subject of the present invention to provide a stable aqueous liquid  
30 composition comprising a cyclodextrin or a cyclodextrine derivative, a resin finishing or cross-linking agent, and at least one emulsifier. The emulsifier aids formulation flexibility when other finishing agents, such as softeners, oil- or water repellants, stain release agents, flame retardants or other effect finishing agents are applied together to the textile in one finishing step. The resin finishing or crosslinking agent aids to bind the cyclodextrine or its derivative

to the material, especially to the textile fiber material. A thermal after treatment builds a polymeric film on the material, especially on the textile compound or it causes crosslinking of reactive groups within the material, especially within the textile fiber material at ambient or elevated temperatures. The resin finishing or crosslinking agent is a compound able to

- 5 reticulate to a polymeric film or to undergo chemical reactions with nucleophilic or electrophilic sites or chemical groups of said material. It is surprising and could not be foreseen that the present composition results in a stable finishing liquor.

Any conventionally usable emulsifier can be used for the compositions. Emulsifier systems

- 10 may comprise for example: carboxylic acids and their salts: alcalin soap of sodium, potassium and ammonium, metallic soap of calcium or magnesium, organic basis soap such as Lauric, palmitic, stearic and oleic acid etc... Alkyl phosphates or phosphoric acid esters, acid phosphate, diethanolamine phosphate, potassium cetyl phosphate. Ethoxylated carboxylic acids or polyethyleneglycol esters, (Polyethylene glycol) PEG-n acylates. Linear

- 15 fatty alcohols having from 8 to 22 carbon atoms, branched from 2 to 30 mol of ethylene oxide and/or from 0 to 5 mol propylene oxide with with fatty acids having from 12 to 22 carbon atoms and with alkylphenols having from 8 to 15 carbon atoms in the alkyl group. Fatty alcohol polyglycoether such as laureth-n, ceteareth-n, steareth-n, oleth-n. Fatty acid polyglycoether such as PEG-n stearate, PEG-n oleate, PEG-n cocoate. Monoglycerides and

- 20 polyol esters. C12-C22 fatty acid mono- and di-esters of addition products of from 1 to 30 mol of ethylene oxide with polyols. Fatty acid and polyglycerol ester such as monostearate glycerol, diisostearoyl polyglyceryl-3-diisostearates, polyglyceryl-3-diisostearates, triglyceryl diisostearates, polyglyceryl-2-sesquiisostearates or polyglyceryl dimerates. Mixtures of compounds from a plurality of those substance classes are also suitable. Fatty acid

- 25 polyglycoesters such as monostearate diethylene glycol, fatty acid and polyethylene glycol esters, fatty acid and saccharose esters such as sucro esters, glycerol and saccharose esters such as sucro glycerides. Sorbitol and sorbitan, sorbitan mono- and di-esters of saturated and unsaturated fatty acids having from 6 to 22 carbon atoms and ethylene oxide addition products. Polysorbate-n series, sorbitan esters such as sesquiisostearate, sorbitan,

- 30 PEG-(6)-isostearate sorbitan, PEG-(10)-sorbitan laurate, PEG-17- dioleate sorbitan. Glucose derivatives, C8-C22 alkyl-mono and oligo-glycosides and ethoxylated analogues with glucose being preferred as the sugar component. O/W emulsifiers such as methyl gluceth-20 sesquistearate, sorbitan stearate/sucrose cocoate, methyl glucose sesquistearate, cetearyl alcohol/cetearyl glucoside. W/O emulsifiers such as methyl glucose dioleate/ methyl glucose

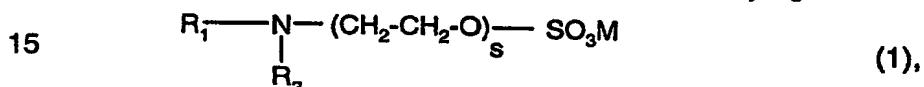
isostearate. Sulfates and sulfonated derivatives, dialkylsulfosuccinates, dioctyl succinate, alkyl lauryl sulfonate, linear sulfonated parafins, sulfonated tetrapropyne sulfonate, sodium lauryl sulfates, ammonium and ethanolamine lauryl sulfates, lauryl ether sulfates, sodium laureth sulfates, sulfosuccinates, aceyl isothionates, alkanolamide sulfates, taurines, methyl 5 taurines, imidazole sulfates. Amine derivatives, amine salts, ethoxylated amines, oxide amine with chains containing an heterocycle such as alkyl imidazolines, pyridine derivatives, isoquinoline, cetyl pyridinium chloride, cetyl pyridinium bromide, quaternary ammonium such as cetyltrimethylbromide ammonium bromide (CTBA), stearylalkonium. Amide derivatives, 10 alkanolamides such as acylamide DEA, ethoxylated amides such as PEG-n acylamide, oxydeamide. Polysiloxane/polyalkyl/polyether copolymers and derivatives, dimethicone, copolyols, silicone polyethylene oxide copolymer, silicone glycol copolymer. Propoxylated or POE-n ethers (Meroxapols), Polaxamers or poly(oxyethylene)m-block-poly(oxypropylene)n-block(oxyethylene). Zwitterionic surfactants that carry at least one quaternary ammonium group and at least one carboxylate and/or sulfonate group in the molecule. Zwitterionic 15 surfactants that are especially suitable are betaines, such as N-alkyl-N,N-dimethylammonium glycinate, cocoalkyldimethylammonium glycinate, N-acylamino-propyl-N,N-dimethylammonium glycinate, cocoacylaminopropylidimethylammonium glycinate and 2-alkyl-3-carboxymethyl-3-hydroxyethylimidazolines each having from 8 to 18 carbon atoms in the alkyl or acyl group and also cocoacylaminooethylhydroxyethylcarboxy- 20 methylglycinate, N-alkylbetaine, N-alkylaminobetaines. Alkylimidazolines, alkylopeptides, lipopainoacides, self emulsifying bases and the compounds as described in K.F.DePolo, A short textbook of cosmetology, Chapter 8, Table 8-7, p250-251.

Non ionic bases such as PEG-6 beeswax (and) PEG-6 stearate (and) polyglyceryl -2-isostearate [Apifac], glyceryl stearate ( and) PEG-100 stearate. [Arlacel 165], PEG-5 glyceryl stearate [arlatone 983 S], sorbitan oleate (and) polyglyceryl-3 ricinoleate.[Arlacel 1689], sorbitan stearate and sucrose cocoate [arlatone 2121], glyceryl stearate and laureth-23 [Cerasynth 945], cetearyl alcohol and ceteth-20 [Cetomacrogol Wax], cetearyl alcohol and colysorbate 60 and PEG-150 and stearate-20[Polawax GP 200, Polawax NF], cetearyl 25 alcohol and cetearyl polyglucoside [Emulgade PL 1618], cetearyl alcohol and ceteareth-20 [Emulgade 1000NI, Cosmowax], cetearyl alcohol and PEG-40 castor oil [Emulgade F Special], cetearyl alcohol and PEG-40 castor oil and sodium cetearyl sulfate [Emulgade F], stearyl alcohol and steareth-7 and steareth-10 [Emulgator E 2155], cetearyl alcohol and szearth-7 and steareth-10 [Emulsifying wax U.S.N.F], glyceryl stearate and PEG-75

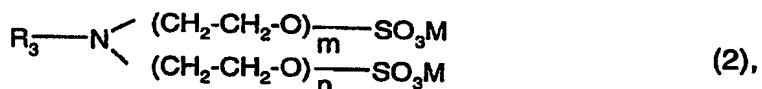
- stearate [Gelot 64], propylene glycol ceteth-3 acetate .[Heteester PCS], propylene glycol isoceth-3 acetate [Heteester PHA], cetearyl alcohol and ceteth-12 and oleth-12 [Lanbritol Wax N 21], PEG -6 stearate and PEG-32 stearate [Tefose 1500], PEG-6 stearate and ceteth-20 and steareth-20 [Tefose 2000], PEG-6 stearate and ceteth-20 and glyceryl stearate and  
5 steareth-20 [Tefose 2561], glyceryl stearate and ceteareth-20 [Teginacid H, C, X].

- Anionic alkalin bases such as PEG-2 stearate SE, glyceryl stearate SE [Monegine, Cutina KD], propylene glycol stearate [Tegin P]. Anionic acid bases such as cetearyl Alcohol and Sodium cetearyl sulfate [Lanette N, Cutina LE, Crodocol GP], cetearyl alcohol and sodium  
10 lauryl sulfate [Lanette W], trilaneth-4 phopshate and glycol stearate and PEG-2 stearate [Sedefos 75], glyceryl stearate and sodium lauryl Sulfate [Teginacid Special]. Cationic acid bases such as cetearyl alcohol and cetrimonium bromide.

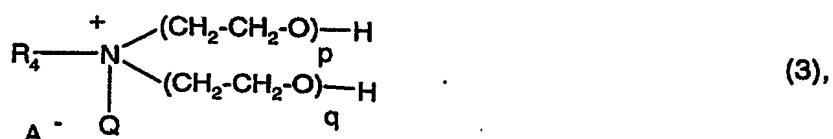
Preferred emulsifiers contain at least one auxiliary agent of the formula



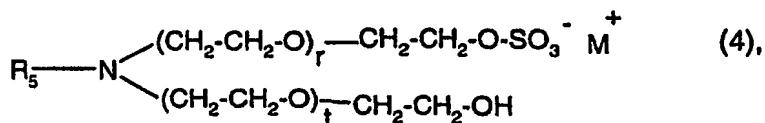
wherein  $R_1$  and  $R_2$  is alkyl or alkenyl having 12 bis 24 carbon atoms, M is hydrogen, alkali metal or ammonium und s is an integer of 2 to 14,



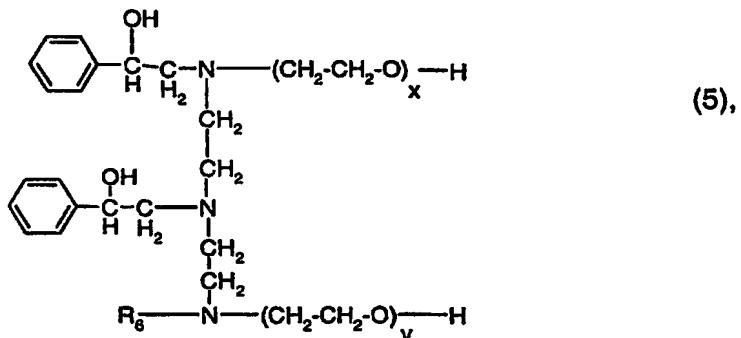
- 20 wherein  $R_3$  is alkyl or alkenyl having 12 bis 24 carbon atoms, M is hydrogen, alkali metal or ammonium and m und n are integers such that the sum of m and n is 2 to 14,



- wherein  $R_4$  is alkyl or alkenyl having 12 to 24 carbon atoms, Q is  $C_1-C_4$  alkyl, A is an anion,  
25 especially the  $CH_3-SO_4$ -Anion and p und q are integers such that the sum of p and q is 15 to 55,

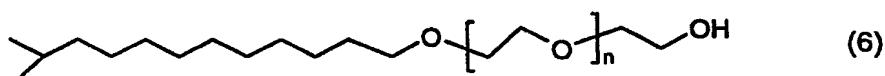


wherein R<sub>5</sub> is alkyl or alkenyl having 12 to 24 carbon atoms, r and t are integers such that the sum of r and t is 14 to 19 and M is an alkali metal or ammonium.



- 5 R<sub>6</sub> is alkyl or alkenyl having 12 to 22 carbon atoms, x and y are integers such that the sum of x and y is 80 to 140, and

isotridecylalcohol containing 6 to 15 mols ethylene oxide of the formula



wherein n is an integer from 6 to 15.

10

The radical Q and the anion A<sup>-</sup> in formula (3) are derived from quaternising agents. Q being a substituted or unsubstituted alkyl radical. Examples of suitable such quaternising agents are chloroacetamide, ethyl bromide, ethylenechlorohydrine, ethylenebromohydrine, epichlorohydrine, epibromohydrine, and, in particular, dimethyl sulfate.

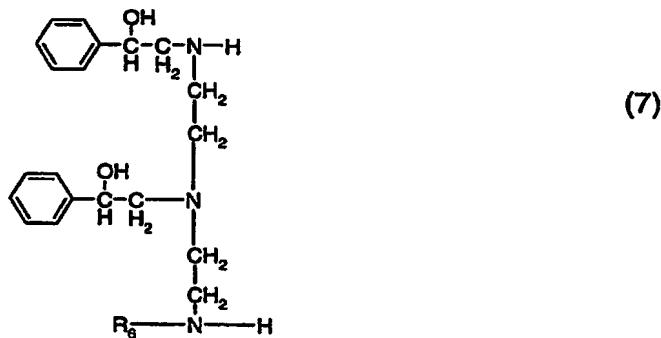
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**Especially preferred emulsifiers or mixtures thereof are those of the formula (3), (4) and (6).**

The emulsifiers may be used in an amount, wherein the molar ratio of cyclodextrin or cyclodextrin derivative and emulsifier is 1 : 0.005 to 1 : 10, preferred is a molar ratio of cyclodextrine or cyclodextrine derivative and emulsifier of 1 : 0.05 to 1 : 2, an especially preferred molar ratio of cyclodextrine or cyclodextrine derivative and emulsifier is 1 : 0.2 to 1 : 1.

The compounds of the formulae (1), (2), (3), (4), (5) and (6) are known and can be prepared according to known processes. Compounds of formula (1) and (2) can be prepared by

- addition of 2 to 14 mols ethylene oxide onto aliphatic amines which have an alkyl or alkenyl radical having 12 to 24 carbon atoms, and converting the adduct into the acid ester and the latter, if desired, into its alkali metal or ammonium salts. Compounds of formula (3) are prepared by addition of, for example, 15 to 55 mols of ethylene oxide onto aliphatic amines
- 5 which have an alkyl or alkenyl radical havin 12 to 24 carbon atoms, and reacting the adduct with one of the above mentioned quaternising agents to give the compound of the formula (3). Compounds of the formula (4) are prepared in analogy to the compounds of the formula (1) using a smaller amount to convert the adduct into the acid ester and the latter, if desired, into its alkali metal or ammonium salts. Compounds of the formula (5) are prepared by the
- 10 addition of 80 to 140 mols of ethylene oxide onto a compound of the formula



in which R<sub>6</sub> is as defined under formula (5).

- Amines required as starting materials in the preparation of the compounbds of the formulae
- 15 (1), (2), (3) and (4) can have saturated or unsaturated, branched or unbranched hydrocarbon radicals having 12 to 24, preferably 16 to 22 carbon atoms. The amines can be single compounds or be in the form of mixtures. The amine mixture used are preferably those formed in the conversion of natural fats or oils, for example tallow fat or soya bean or coconut oil, into the corresponding amines. Specific examples of amines are dodecylamine,
- 20 hexadecylamine, octadecylamine, arachidylamine, behenylamine, and octadecenylamine. Not only the addiction of ethylene oxide but also the esterification can be carried out according to methods known per se. The sterification can be performed with sulfuric acid or its functional derivatives, for example chlorosulfonic acid or, in particular, sulfamic acid. The esterification is generally carried out by simply mixing the reactants while heating them,
- 25 advantageously at a temperature between 50° and 100°C. The free acids can be converted into the alkali metal or ammonium salts by adding in a conventional manner bases, for example moonia or sodium or potassium hydroxide.

The cyclodextrin derivatives may be uncomplexed or complexed, for example, with antimicrobials, biocides, bactericides, insecticides, fungicides, pharmaceutical active compounds, UV-stabilizers, perfumes, fragrances, pheromones, vitamines or skin-, hair and textile benefit agents, e.g. UV-absorber, fatty acids, anti-irritants, inflammatory agents or

- 5 cosmetic agents, e.g. skin firming, soothing, warming, cooling, wrinkle reducing or control agents.

The preparation of complexed reactive cyclodextrine derivatives can be carried out according to known processes, such as described in WO-A-02/22941.

10

As cyclodextrine derivatives it is suitable to use hydroxypropyl- $\beta$ -cyclodextrine, other non reactive substituted  $\beta$ -cyclodextrine, or a cyclodextrine having a fibre reactive group.

15

Reactive groups of the cyclodextrin derivatives are groups capable of reacting with functional groups of a suitable substrate, such as a textile fibre material, for example with the hydroxyl groups of cellulose, the amino, carboxyl, hydroxyl or thiol groups in the case of wool and silk or with the amino and possibly carboxyl groups of synthetic polyamides with the formation of covalent chemical bonds. Reactive groups are generally attached directly or via a bridge member to a carbon atom of the cyclodextrin derivative. Examples of suitable reactive groups include those, which contain at least one detachable substituent on an aliphatic, aromatic or preferably on a heterocyclic radical or in which the radicals mentioned contain a radical suitable for reaction with the fibre material. Examples of suitable bridge members according to which the fibre-reactive groups can be attached to a carbon atom of the cyclodextrin derivative are -NH-, -O-CO- and -O-, especially -O-CO- and -O-, and preferably -O-

25

Reactive cyclodextrin derivatives are known and the preparation of such reactive cyclodextrin derivatives can be carried out according to known processes.

30

Reactive cyclodextrin derivatives containing aldehyde groups capable of reacting with the hydroxyl groups of cellulose or polyvinyl alcohol are described in EP-A-0 483 380.

A. Deratani, B. Pöpping, Makromol. Chem. Rapid Commun. 13 (1992) 237 describe the preparation of cyclodextrin chlorohydrin (3-chloro-2-hydroxypropyl-cyclodextrin derivative) by reaction of  $\alpha$ -cyclodextrin in an aqueous medium with epichlorohydrin.

German patent application No. 101 55 782.5 furthermore discloses the preparation of  $\alpha$ -cyclodextrin derivatives containing reactive 2,3-dibromopropionyl- or vinylsulfonyl groups which are well known reactive anchor groups in the field of reactive dyestuffs.

5

Basically, all fibre reactive groups known in the field of reactive dyestuffs, such as described in Venkataraman "The Chemistry of Synthetic Dyes" Vol. 6, pages 1-209, Academic Press, New York, London 1972 or EP-A-625 549 and US-A-5 684 138 are suitable.

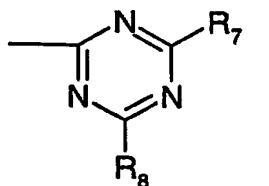
- 10 Further examples of reactive groups include reactive radicals containing carbo- or heterocyclic 4-, 5- or 6-rings substituted by a detachable atom or group. Examples of heterocyclic radicals include heterocyclic radicals which contain at least one detachable substituent attached to a heterocyclic ring; and those which contain at least one reactive substituent attached to a 5- or 6-membered heterocyclic ring, as to a triazine, pyridine or  
15 pyrimidine. The heterocyclic reactive radicals mentioned may further contain, via a direct bond or via a bridge member, further reactive radicals. Such reactive cyclodextrin derivatives are described in US-A-5 728 823.

- 20 The present invention also includes compositions containing the hydrolyzate of the reactive cyclodextrin derivative. Hydrolyzates of reactive cyclodextrin derivatives are formed upon the reaction of the reactive groups with water in the known manner. Compositions containing reactive cyclodextrin derivatives are preferred.

- 25 Preferably the reactive group of the cyclodextrin derivative is vinylsulfonyl,  $\alpha,\beta$ -dihalopropionyl,  $\alpha$ -haloacryloyl, wherein halo is e.g. bromo or chloro, in particular bromo, or a nitrogen-containing heterocycle having at least one substituent selected from the group consisting of halogen, especially fluorine or chlorine, and unsubstituted or substituted pyridinium. It is to be understood that vinylsulfonyl will also include the precursors thereof, which correspond to the formula  $-\text{SO}_2\text{-CH}_2\text{-CH}_2\text{-Z}$ , wherein Z is a group removable under alkaline conditions. Z is for example -Cl, -Br, -F, -OSO<sub>3</sub>H, -SSO<sub>3</sub>H, -OCO-CH<sub>3</sub>, -OPO<sub>3</sub>H<sub>2</sub>, -OCO-C<sub>6</sub>H<sub>5</sub>, -OSO<sub>2</sub>-C<sub>1</sub>-C<sub>4</sub>alkyl or -OSO<sub>2</sub>-N(C<sub>1</sub>-C<sub>4</sub>alkyl)<sub>2</sub>. Z is preferably a group of formula -Cl, -OSO<sub>3</sub>H, -SSO<sub>3</sub>H, -OCO-CH<sub>3</sub>, -OCO-C<sub>6</sub>H<sub>5</sub> or -OPO<sub>3</sub>H<sub>2</sub>, especially -Cl or -OSO<sub>3</sub>H and more especially -OSO<sub>3</sub>H.

More preferably the reactive group of the cyclodextrin derivative is a nitrogen-containing heterocycle having at least one substituent selected from the group consisting of halogen, especially fluorine or chlorine, and unsubstituted or substituted pyridinium.

- 5 Specific examples of nitrogen-containing heterocyclic reactive groups are  
a) a triazine group of formula



wherein

R<sub>7</sub> is fluorine, chlorine, unsubstituted or carboxy-substituted pyridinium or hydroxy, preferably

10 fluorine, chlorine or unsubstituted or carboxy-substituted pyridinium, and

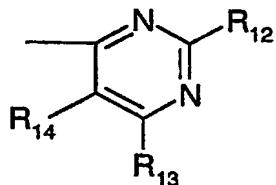
R<sub>8</sub> is as defined above for R<sub>7</sub> or is a radical of formula -OR<sub>9</sub> or -N(R<sub>10</sub>)R<sub>11</sub>, wherein

R<sub>9</sub> is hydrogen, alkali, C<sub>1</sub>-C<sub>8</sub>alkyl which is unsubstituted or substituted by hydroxy or C<sub>1</sub>-C<sub>4</sub>alkoxy, and

R<sub>10</sub> and R<sub>11</sub>, independently from each other, are hydrogen; C<sub>1</sub>-C<sub>8</sub>alkyl which is unsubstituted

15 or substituted by C<sub>1</sub>-C<sub>4</sub>alkoxy, hydroxy, sulfo, sulfato or carboxy; or phenyl which is unsubstituted or substituted by C<sub>1</sub>-C<sub>4</sub>alkyl, C<sub>1</sub>-C<sub>4</sub>alkoxy, halogen, nitro, carboxy or sulfo;

b) or a pyrimidinyl group of formula

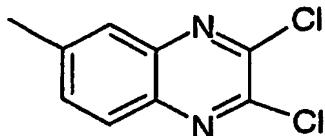


wherein

20 one of radicals R<sub>12</sub> and R<sub>13</sub> is fluorine or chlorine and the other one of radicals R<sub>12</sub> and R<sub>13</sub> is fluorine, chlorine, or is a radical of formula -OR<sub>9</sub> or -N(R<sub>10</sub>)R<sub>11</sub> as defined above, and

R<sub>14</sub> is C<sub>1</sub>-C<sub>4</sub>alkylsulfonyl, C<sub>1</sub>-C<sub>4</sub>alkoxysulfonyl, C<sub>1</sub>-C<sub>4</sub>alkoxycarbonyl, C<sub>2</sub>-C<sub>4</sub>alkanoyl, chlorine, nitro, cyano, carboxyl or hydroxyl;

c) or a dichloroquinoxaline group of formula



(10).

Highly preferred are groups of formula (6), especially those wherein R<sub>7</sub> is chlorine.

R<sub>8</sub> is preferably a radical of formula -OR<sub>9</sub>, wherein R<sub>9</sub> is hydrogen, alkali or C<sub>1</sub>-C<sub>8</sub>alkyl.

- 5 Preferred radicals R<sub>8</sub> are those of formula -OR<sub>9</sub> wherein R<sub>9</sub> is hydrogen, alkali or C<sub>1</sub>-C<sub>4</sub>alkyl, especially hydrogen or alkali. Alkali is highly preferred. Alkali is preferably sodium.

The cyclodextrin derivatives contain e.g. 1 to 4, preferably 2 to 3 reactive groups.

- 10 The cyclodextrin derivatives may be uncomplexed or complexed, for example, with antimicrobials, biocides, bactericides, insecticides, fungicides, pharmaceutical active compounds, UV-stabilizers, perfumes, fragrances, pheromones, vitamines or skin-, hair and textile benefit agents, e.g. UV-absorber, fatty acids, anti-irritants or inflammatory agents or cosmetic agents, e.g. skin firming, soothing, warming, cooling, wrinkle reducing or control agents.
- 15

The preparation of complexed reactive cyclodextrine derivatives can be carried out according to known processes, such as described in WO-A-02/22941.

- 20 The compositions may also comprise pH regulators, such as mineral acids, organic acids or bases.

As further additives, the aqueous liquid compositions according to the present invention may comprise textile-finishing agents, humectants, defoaming agents and antifreezing agents.

25

As textile finishing agents in the compostions according to the invention come into consideratation textile chemicals or compositions with the aim of adding a new or improving textile performance or consumer acceptability or desirability such as softening, raising, sanforizing, oil or water repellents, antifibrillation, flame retardant agents.

30

As humectants in the compositions according to the invention there come into consideration, for example, urea or a mixture of sodium lactate (advantageously in the form of a 50 % to 60 % aqueous solution) and glycerol and/or propylene glycol in amounts of preferably from 0.1 to 30 % by weight, especially from 2 to 30 % by weight.

5

The aqueous liquid composition can be prepared, for example, in the following manner.  $\beta$ -cyclodextrine, hydroxypropyl- $\beta$ -cyclodextrine or one of the fiber reactive cyclodextrines (e.g. Cavasol W7 MCT<sup>®</sup>) is mixed with water and an emulsifier or a mixture of two or more emulsifiers, especially in a molar amount of cyclodextrine : emulsifier from 1:0.005 to 1:10.

10

Thereafter, water and a resin finishing or crosslinking agent is added, and if desired other textile finishing agents are added to the solution to obtain an aqueous liquid composition having the desired concentration of the cyclodextrin in textile finishing processes.

15

The amount of the resin finishing or crosslinking agent in the composition may vary between 1 and 300 gram per litre. The amount of cyclodextrine or cyclodextrine derivative may vary between 0,5 and 100 gram per litre.

20

The compositions according to the invention are suitable as benefit-agents in a finishing process for widely varying kinds of substrates, such as paper, textile fibre materials, keratinous fibres, e.g. human hair, leather, or films or foils of plastic made of polymers susceptible to the reaction with the fibre reactive cyclodextrin, for example, polyvinylalcohol or polymers or copolymers of acrylic acid, methacrylic acid or hydroxyethylmethacrylate, thereby modifying the surface of such substrates.

25

As resin finishing or crosslinking agent in a textile finishing process which may result in superior effect durability different agents can be used, for example, dimethyol-urea, dimethoxy-methyl-urea, trimethoxy-methyl-melamin, tetramethoxy-methyl-melamine, hexamethoxy-methyl-melamine, dimethyol-dihydroxy-ethylene-urea, dimethyol-propylene-urea, dimethyol-4-methoxy-5,5'-dimethyl-propylene-urea, dimethyol-5-hydroxypropylene-urea, butane-tetra-carboxylic-acid, citric acid, maleic acid, bonding agents, especially acrylates, silicones, urethanes and butadienes. Such textile finishing processes are described, for example, in DE-A-40 35 378.

A group of preferred resin finishing or crosslinking agents is selected from the group consisting of dimethylol-urea, dimethoxy-methyl-urea, trimethoxy-methyl-melamin, hexamethoxy-methyl-melamine, dimethylol-dihydroxy-ethylene-urea, dimethylol-propylene-urea, dimethylol-4-methoxy-5,5'-dimethyl-propylene-urea, dimethylol-5-hydroxypropylene-urea, butane-tetra-carboxylic-acid.

A very preferred resin finishing or crosslinking agent is dimethylol-dihydroxyethylene-urea.

The concentration of the resin finishing or crosslinking agent on the material, especially the textile fiber material is between 0.1 and 30% by weight of the weight of the material.

In order to achieve a two- or three-dimensional crosslinking and the connection of the cyclodextrine or its derivative with the material, especially the textile fiber material it is preferred to heat the material in an after treatment step up to 100 to 230°C for 20 seconds to 10 minutes. Alternatively treatments of said material is also possible at ambient to elevated temperatures in the moist or wet material state; these are usually carried out under acidic conditions.

Preferred, as the substrates are textile fibre materials containing hydroxyl groups or containing nitrogen. Textile fibre materials can be in the form of fibre, yarn or piece goods, such as non-wovens, knitted and woven goods, pile fabrics or terry goods. Examples are silk, wool, polyamide fibres and polyurethanes, and in particular all types of cellulosic fibre materials. Such cellulosic fibre materials are, for example, the natural cellulosic fibres, such as cotton, linen and hemp, as well as cellulose and regenerated cellulose. The compositions according to the invention are also suitable for finishing fibres containing hydroxyl groups, which are contained in blend fabrics, for example mixtures of cotton with polyester fibres or polyamide fibres. The compositions according to the invention are particularly suitable for finishing cellulosic materials. They can furthermore be used for finishing natural or synthetic polyamide fibre materials.

The compositions according to the invention are applied to a finishing process which is a further subject of the present invention. The cyclodextrins are fixed, if appropriate after an alkali treatment, under the action of heat, steam or by storage at room temperature or slightly elevated temperature for several hours, thereby forming a durable cyclodextrine finish.

The finished substrates contain, for example, 0.05 to 25% by weight, preferably 0.1 to 10% by weight, and most preferably 0.5 to 3% by weight, of the cyclodextrine derivative, based on the total weight of the substrate.

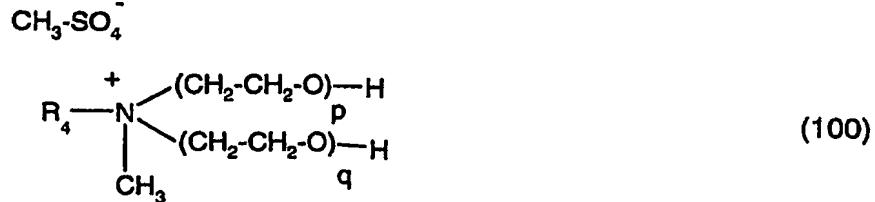
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The finished substrates can be used to complex within the cyclodextrin cavity, for example, UV-stabilizers, antimicrobials, biocides, bactericides, insecticides, fungicides, pharmaceutical active compounds, fragrances, perfumes, pheromones, vitamines or skin-, hair and textile benefit agents, e.g. UV-absorber, fatty acids, anti-irritants or inflammatory agents, or

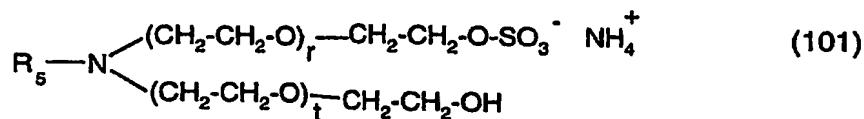
- 10 cosmetic agents, to e.g. solubilize water-insoluble or poorly water-soluble substances, to increase the bioavailability of active compounds; to stabilize substances against light, temperature, oxidation, hydrolysis or from volatility, to mask bad taste or unpleasant odor, to slowly release active compounds in a controlled manner over a prolonged period of time (delivery systems). On the other side, the finished substrates are useful to assimilate  
15 chemical substances, e.g. from a gaseous or liquid environment, which are captured in the cyclodextrin cavity, thereby serving as a collector system. Such collector systems may find application in the field of medical diagnostics or help to determine pollutants from the environment. Decomposition products of perspiration are trapped in the cyclodextrin cavity, thus diminishing or preventing malodor. Textile materials, such as clothings finished with the  
20 inventive composition stay fresh with a pleasant smell.

The inventive compositions are distinguished by good stability of the final finishing formulation for the textile treatment improving process reliability and finishing quality..

- 25 The following examples serve to illustrate the invention. Temperatures are stated in degrees Celsius, parts are parts by weight and the percentage data are based on percentages by weight, unless noted otherwise. Parts by weight bear the same relation to parts by volume as the kilogram to the litre.
- 30 Example 1: Cavasol® W7 HP TL (Hydroxypropyl- $\beta$ -cyclodextrine available from Wacker Chemie AG, Germany) is neutralized. An aqueous composition is prepared by mixing the following components: 45 parts per litre of neutralized Cavasol® W7 HP TL and 5 parts per litre of an aqueous solution containing 25 % by weight of water, 25 % by weight of the compound of the formula



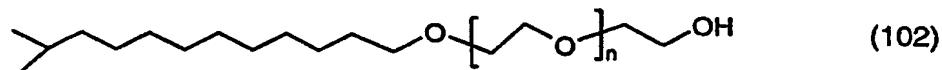
wherein  $\text{R}_4$  is  $\text{CH}_3(\text{CH}_2)_{17-21}-$ , and the sum of  $p$  and  $q$  is 34, and 50% by weight of the compound of the formula



- 5 wherein  $\text{R}_5$  is  $\text{CH}_3(\text{CH}_2)_{15-18}-$  and  $\text{CH}_3(\text{CH}_2)_7\text{-CH=CH-(CH}_2)_8-$  and the sum of  $r$  and  $t$  is 14, are mixed together with 40 parts per litre of CIBA Knittex® FEL, 12 parts per litre of magnesium chloride and 30 parts per litre of Ciba Ultraphil® HMS (textile finishing agent available from Ciba Specialty Chemicals) to give a stable composition without any flocculation.

10

- Example 2: Cavasol® W7 HP TL (Hydroxypropyl- $\beta$ -cyclodextrine available from Wacker Chemie AG, Germany) is neutralized. An aqueous composition is prepared by mixing the following components: 47.5 parts per litre of neutralized Cavasol® W7 HP TL and 2.75 parts per litre of an aqueous solution containing 50% by weight of water and 50% by weight of 15 the isotridecyl alcohol of the formula



- wherein  $n$  is 7 or 9, are mixed together with 40 parts per litre of CIBA Knittex® FEL, 12 parts per litre of magnesium chloride and 30 parts per litre of Ciba Ultraphil® HMS (textile finishing agent available from Ciba Specialty Chemicals) to give a stable composition without any 20 flocculation.

- Application Example 1: 1000 parts of the composition of Example 1 is used. The pH value is adjusted to 4.5 to 5. Single cotton Jersey is impregnated with the composition such that it increases by 80% of its weight, subsequently dried at room temperature ( $25^\circ\text{C}$ ) and fixed for 25 40 seconds at  $170^\circ\text{C}$ .

The finished fabrics are submitted to a washing test at 60°C (0 time, 5 times, 10 times and 20 times) under the following conditions:

Detergent: 30 g IEC 456-A (standard detergent)

Washing machine: Wascator FOM 71MP LAB.

5

The presence of cyclodextrine on the fabric was verified by the decoloration of an alkaline phenolphthaleine solution. Ability of the finished fabric to reduce malodor has been tested by keeping 1 g specimens of the finished fabrics (unwashed, 5 x washed, 10 x washed and 20 x washed) over night in closed bottles (content 100 ml) containing 50 mg of water and 20 mg

10 of an ethanol solution (80 vol.-%) containing 0.1 % p-cresol. The specimens were air-ventilated for 5 minutes. In any case a distinct reduction of malodor (p-cresol) was detected by 4 different test persons compared to unfinished specimens treated in the same manner.

Application Example 2: 1000 parts of the composition of Example 2 is used. Different fabrics

15 (knitted cotton, viscose and cotton/polyester (50/50) blends) are impregnated with the solution such that it increases by 70% of its weight, subsequently dried at room temperature (25°C) and fixed for 1 minute at 160°C.

20 The finished fabrics are submitted to a washing test at 60°C (0 time, 5 times, 10 times and 20 times) under the following conditions:

Detergent: 30 g IEC 456-A (standard detergent)

Washing machine: Wascator FOM 71MP LAB.

25 The presence of cyclodextrine on the fabric was verified by the decoloration of an alkaline phenolphthaleine solution. Ability of the finished fabric to reduce malodor has been tested by keeping 1 g specimens of the finished fabrics (unwashed, 5 x washed, 10 x washed and 20 x washed) over night in closed bottles (content 100 ml) containing 50 mg of water and 20 mg of an ethanol solution (80 vol.-%) containing 0.1 % p-cresol. The specimens were air-ventilated for 5 minutes. In any case a distinct reduction of malodor (p-cresol) was detected by 4 different test persons compared to unfinished specimens treated in the same manner.

Claims

1. An aqueous liquid composition comprising

- 5      a) a cyclodextrin or a derivative thereof,
- b) a resin finishing or crosslinking agent, and
- c) at least one emulsifier.

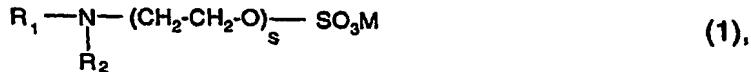
10     2. An aqueous composition according to claim 1, wherein component a) is  $\beta$ -cyclodextrine or hydroxypropyl- $\beta$ -cyclodextrine.

15     3. A composition according to claim 1 or 2, wherein component a) is a reactive cyclodextrin derivative or the hydrolyzate thereof.

20     4. A composition according to any of claims 1 to 3, wherein component a) is present in an amount of 0.05 to 70 % by weight, based on the total weight of the composition.

25     5. A composition according to any of claims 1 to 4, wherein the molar ratio of cyclodextrin or cyclodextrin derivative and emulsifier is 1 : 0.005 to 1 : 10, preferred is a molar ratio of cyclodextrine or cyclodextrine derivative and emulsifier of 1 : 0.05 to 1 : 2, an especially preferred molar ratio of cyclodextrine or cyclodextrine derivative and emulsifier is 1 : 0.2 to 1 : 1.

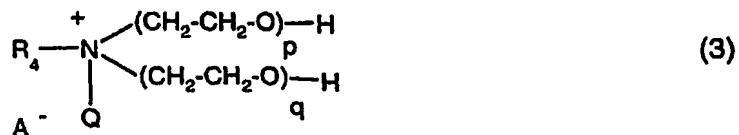
30     6. A composition according to any of claims 1 to 5, wherein as component (c) at least one emulsifier of the formulae (1), (2), (3), (4), (5) and (6) is used



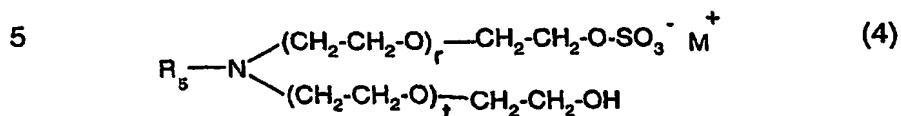
wherein  $R_1$  and  $R_2$  is alkyl or alkenyl having 12 bis 24 carbon atoms, M is hydrogen, alkali metal or ammonium und s is an integer from 2 to 14,



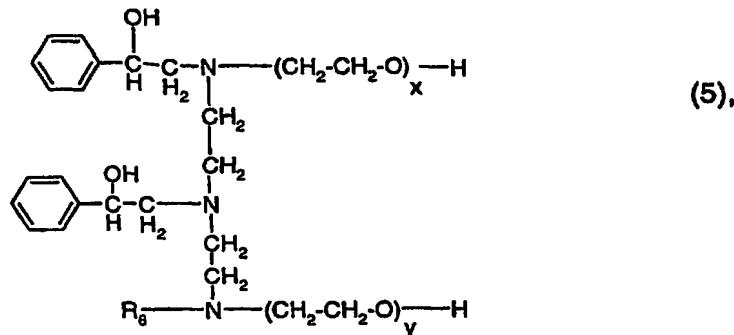
wherein  $R_3$  is alkyl or alkenyl having 12 bis 24 carbon atoms, M is hydrogen, alkali metal or ammonium and m und n are integers such that the sum of m and n is 2 to 14,



wherein R<sub>4</sub> is alkyl or alkenyl having 12 to 24 carbon atoms, Q is C<sub>1</sub>-C<sub>4</sub> alkyl, A is an anion, especially CH<sub>3</sub>-SO<sub>4</sub><sup>-</sup>-Anion and p und q are integers such that the sum of p and q is 15 to 55,



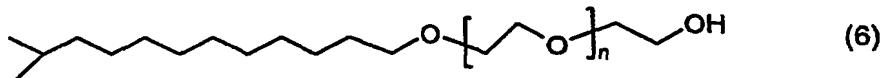
wherein R<sub>5</sub> is alkyl or alkenyl having 12 to 24 carbon atoms, r and t are integers such that the sum of r and t is 14 to 19 and M is an alkali metal or ammonium, or



R<sub>6</sub> is alkyl or alkenyl having 12 to 22 carbon atoms, x and y are integers such that the sum of

10 x and y is 80 to140, and

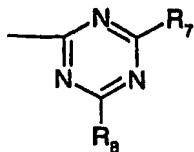
isotridecylalcohol containing 6 to 15 mols ethylene oxide of the formula



wherein n is an integer from 6 to 15.

15 7. A composition according to claim 3, wherein the reactive group of the cyclodextrin derivative is a nitrogen-containing heterocycle having at least one substituent selected from the group consisting of halogen and unsubstituted or substituted pyridinium.

8. A composition according to claim 7, wherein the reactive group of the cyclodextrin  
20 derivative is  
a) a triazine group of formula

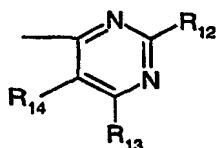


(8)

wherein

R<sub>7</sub> is fluorine, chlorine, unsubstituted or carboxy-substituted pyridinium or hydroxy, and  
R<sub>8</sub> is as defined above for R<sub>7</sub> or is a radical of formula -OR<sub>9</sub> or -N(R<sub>10</sub>)R<sub>11</sub>, wherein

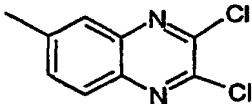
- 5 R<sub>9</sub> is hydrogen, alkali, C<sub>1</sub>-C<sub>8</sub>alkyl which is unsubstituted or substituted by hydroxy or  
C<sub>1</sub>-C<sub>4</sub>alkoxy, and  
R<sub>10</sub> and R<sub>11</sub>, independently from each other, are hydrogen; C<sub>1</sub>-C<sub>8</sub>alkyl which is unsubstituted  
or substituted by C<sub>1</sub>-C<sub>4</sub>alkoxy, hydroxy, sulfo, sulfato or carboxy; or phenyl which is  
unsubstituted or substituted by C<sub>1</sub>-C<sub>4</sub>alkyl, C<sub>1</sub>-C<sub>4</sub>alkoxy, halogen, nitro, carboxy or sulfo; or  
10 b) a pyrimidinyl group of formula



(9)

wherein one of radicals R<sub>12</sub> and R<sub>13</sub> is fluorine or chlorine and the other one of radicals R<sub>12</sub>  
and R<sub>13</sub> is fluorine, chlorine, or is a radical of formula -OR<sub>9</sub> or -N(R<sub>10</sub>)R<sub>11</sub> as defined above,  
and

- 15 R<sub>14</sub> is C<sub>1</sub>-C<sub>4</sub>alkylsulfonyl, C<sub>1</sub>-C<sub>4</sub>alkoxysulfonyl, C<sub>1</sub>-C<sub>4</sub>alkoxycarbonyl, C<sub>2</sub>-C<sub>4</sub>alkanoyl, chlorine,  
nitro, cyano, carboxyl or hydroxyl; or  
c) a dichloroquinoxaline group of formula



(10).

- 20 9. A composition according to claim 8, wherein the reactive group of the cyclodextrin  
derivative is a triazine group of formula (6), wherein  
R<sub>7</sub> is chlorine, and  
R<sub>8</sub> is a radical of formula -OR<sub>9</sub>, wherein R<sub>9</sub> is hydrogen, alkali or C<sub>1</sub>-C<sub>8</sub>alkyl, preferably alkali.
- 25 10. A composition according to any of claims 7 to 9, wherein the reactive cyclodextrin  
derivative contains 1 to 4 reactive groups.

11. A composition according to any of claims 1 to 10, wherein the resin finishing agent or the crosslinking agent is able to build a polymeric film on the textile fiber material or has the ability to react with nucleophilic or electrophilic sites or chemical groups within the textile fiber  
5 material..
12. A composition according to claim 11, wherein the resin finishing or crosslinking agent is selected from the group consisting of dimethylol-urea, dimethoxy-methyl-urea, trimethoxy-  
10 methyl-melamin, tetramethoxy-methyl-melamine, hexamethoxy-methyl-melamine, dimethylol-dihydroxy-ethylene-urea, dimethylol-propylene-urea, dimethylol-4-methoxy-5,5'-dimethyl-  
15 propylene-urea, dimethylol-5-hydroxypropylene-urea, butane-tetra-carboxylic-acid, citric acid, maleic acid, bonding agents, especially acrylates, silicones, urethanes and butadienes.
13. A composition according to any of claims 1 to 12, wherein the composition further  
comprises a buffer selected from the group consisting of borax, borates, phosphates, polyphosphates, oxalates, acetates or citrates, in particular phosphates, acetates or citrates.  
15
14. A finishing process comprising treating a substrate with the composition according to  
claim 1.  
20
15. A finishing process according to claim 14, wherein textile fiber material is used as  
substrate.

TM/4-22906/P1

Abstract

An aqueous liquid composition comprising a cyclodextrin or a cyclodextrine derivative, at least one emulsifier and a resin finishing or crosslinking agent is excellent in storage stability and useful in a finishing process for the treatment of suitable substrates, such as fibre materials.

PCT/EP2004/051162

